

## Logarithms:

The logarithm (log to the base 10 or just log on your calculator) is a mathematical functions used in science. You need to be able to manipulate these functions for many reasons, in particular, pH. **pH = -log[H<sup>+</sup>]**

log (base 10) are essentially short-hand way to express orders of magnitude. For example:

$$\log(1) = \log(10^0) = 0$$

$$\log(10) = \log(10^1) = 1$$

$$\log(100) = \log(10^2) = 2$$

$$\log(10^a) = a$$

Make sure you know how to use your calculator using the logarithm function. Try the mathematical expression to see if you get the correct answer:

$$\log(3.60) = 0.556$$

$$\log(36.0) = 1.556$$

$$\log(360.0) = 2.556$$

(Note that the significant figures shown are correct. If you take the log of a number, the numbers past the decimal point in the log should equal the significant figures in the original number. The first number is just shows the order of magnitude.)

$$\log a + \log b = \log(a \cdot b)$$

$$\log a - \log b = \log\left(\frac{a}{b}\right)$$

$$\log(a^n) = n \log(a)$$

Try these examples:

$$\begin{aligned} \log(3.60 \times 10^7) &= \log(3.60) + \log(10^7) \\ &= 0.556 + 7 \\ &= 7.556 \quad (\text{Prove with your calculator !!!!}) \end{aligned}$$

1. Determine the following WITHOUT using a calculator

- a.  $\log(1 \times 10^7)$
- b.  $\log(1 \times 10^{-7})$
- c.  $-\log(1 \times 10^{-7})$
- d.  $\log(1 \times 10^5)$
- e.  $\log(1 \times 10^{-5})$
- f.  $-\log(1 \times 10^{-5})$

2. Determine the following WITH using a calculator

- g.  $\log(3)$
- h.  $\log(7)$

3. Determine the following WITHOUT using a calculator

- a.  $\log(3 \times 10^7)$
- b.  $\log(3 \times 10^{-7})$
- c.  $-\log(3 \times 10^{-7})$
- d.  $\log(7 \times 10^5)$
- e.  $\log(7 \times 10^{-5})$
- f.  $-\log(7 \times 10^{-5})$

4. Find the following solutions WITHOUT using a calculator.

a)  $\log(1000)$

b)  $\log(5)=0.699$

1. What is  $\log(50)$ ?

2. What is  $\log(500)$ ?

3. What is  $\log(0.5)$ ?

4. What is  $\log(0.05)$ ?

**5. What is the pH of a solution that has a  $[H^+]$  of  $5 \times 10^{-2} M$  ?**

**How would you determine the  $[H^+]$  of a solution with a pH of 7?**

There is also an inverse log function ( $10^x$ )

$$10^1 = 10$$

$$10^2 = 100$$

Try some numbers on your calculator to prove to yourself that the expressions below are valid.

$$10^{0.556} = 3.60$$

$$10^{1.556} = 36.0$$

$$10^{2.556} = 360.0$$

**Problems**

**Which is more acidic? How much more?**

1. A sample of rain with a pH of 5 and a sample of lake water with a pH of 4.

2. A tomato juice sample with a pH of 4.5 and a sample of milk with a pH of 6.5